Chemical dynamics and rare events in soft matter physics

Boris Fačkovec

Wales group Department of Chemistry University of Cambridge

Trinity Mathematical Society Cambridge, 22nd February 2015

Chemical dynamics and rare events

1 /23

Big picture

systems of interest in biology, materials, catalysis... ${\color{black}\bullet}$



2 /23

Outline

- motivation
- from *ab initio* to energy landscapes
- friction
- chemical dynamics and the transition state theory
- molecular simulations and the relaxation path sampling
- future directions



Molecules and reactions

description of system in physics - position x and velocity v

$$\frac{\mathrm{d}\mathbf{x}}{\mathrm{d}t} = \mathbf{f}(\mathbf{x}, \mathbf{v}, t) \qquad \qquad \frac{\mathrm{d}\mathbf{v}}{\mathrm{d}t} = \mathbf{g}(\mathbf{x}, \mathbf{v}, t)$$

- molecules clusters of atoms bound by strong bonds
- system described by identity of species and concentration

$$\frac{\mathrm{d}\mathbf{c}}{\mathrm{d}t} = \mathbf{h}(\mathbf{c}, t)$$

solving rate equations -> chemical kinetics

The first principles

"The underlying physical laws necessary for the mathematical theory of a large part of physics and the whole of chemistry are thus completely known, and the difficulty is only that the exact application of these laws leads to equations much too complicated to be soluble." (Paul Dirac)

$$\psi(x_1^e, x_2^e, ..., x_{3M \times n}^e, x_1^n, x_2^n, ..., x_{3N \times n}^n)$$

$$\psi(x_1^e, x_2^e, ..., x_{3M}^e, x_1^n, x_2^n, ..., x_{3N}^n)$$

$$\psi(x_1^e, x_2^e, ..., x_{3M}^e; x_1^n, x_2^n, ..., x_{3N}^n)$$

- separation of molecules
- separation of motion nuclei static compared to electrons

Chemical dynamics and rare events

5 /23

Energy landscapes

 $V(x_1^n, x_2^n, ..., x_{3N}^n)$

- all properties defined by the potential energy function
- in 2D resembles mountain landscape







bf269@cam.ac.uk

Chemical dynamics and rare events

6 /23

Energy, temperature and entropy

- probability of each configuration given by its potential energy
- grouping configurations into observable states results in information loss -> entropy
- grouping of coin tossing, volume

 $P(\mathbf{x}) \propto e^{-\frac{V(\mathbf{x})}{k_B T}}$





Dynamics on energy landscapes



 Can be discretised and numerically propagated from an initial structure = molecular dynamics

Chemical dynamics and rare events

8 /23

Rare events in nature

- ratio of the largest and lowest relevant timescales >> 1
- challenge for molecular dynamics



Density of particles

- in a beaker, we have an ensemble of (10²³) molecules following the same rules
- in the limit of infinitely many particles we can write density
- Newton equation -> Liouville's equation for the density

(in 1D, zero potential)

$$\frac{\partial \rho(x,t)}{\partial t} = -v \frac{\partial \rho(x,t)}{\partial x}$$

$$\frac{\partial \rho(x,v,t)}{\partial t} = \frac{1}{m} \frac{\partial V(x)}{\partial x} \frac{\partial \rho(x,v,t)}{\partial v} - v \frac{\partial \rho(x,v,t)}{\partial x}$$
Chemical dynamics and rare events 10/23 bf269@cam.ac.uk

Adding friction

- motion on a rough landscape
- what is not included in our model causes random bumps

$$m\frac{\mathrm{d}^2 x}{\mathrm{d}t^2} = -\frac{\mathrm{d}V}{\mathrm{d}x} - \gamma v + \sqrt{2k_B T}\gamma^{1/2}\eta(t),$$

Generally:

$$\frac{\partial \rho(\mathbf{x}, \mathbf{v}, t)}{\partial t} = \mathcal{F}\left[\rho(\mathbf{x}, \mathbf{v}, t)\right]$$

no frictionsome frictionlarge frictionSODENewtonLangevinBrownian motionPDELiouvilleFokker-PlanckSmoluchowski

Chemical dynamics and rare events

11/23

Partitioning the space in cells





rate matrix R is calculated from simulations

- $\frac{\mathrm{d}\mathbf{c}}{\mathrm{d}t} = \mathbf{R} \ \mathbf{c}$
- extraction of a smaller system = isolation of cells

General solution to the FPE

$$\frac{\partial \rho(\mathbf{x}, \mathbf{v}, t)}{\partial t} = \mathcal{F}\left[\rho(\mathbf{x}, \mathbf{v}, t)\right]$$

eigenfunctions of the operator can be found only for some systems



The transition state theory

$$\frac{\mathrm{d}}{\mathrm{d}t}[A] = k_{AB}[A]$$

$$k_{\rm AB}^{\rm TST} = \frac{\rm eq.flux}{\rm eq.population}$$

$$k_{\rm a}^{\dagger}(T) = \frac{kT}{h} \frac{Z^{\dagger}}{Z_{\rm a}} e^{-\Delta V/kT}$$

$$k_{\rm a}^{\dagger}(E) = \frac{\bar{\nu}_{\rm a}^{\kappa}}{\bar{\nu}^{\dagger(\kappa-1)}} \left(\frac{E-V^{\dagger}}{E-V_{\rm a}}\right)^{\kappa-1}$$



14/23

Rate constants from trajectories

- (if we have to simulate trajectories) CAN WE DO BETTER?
- for system of 2 states, rate constant from mean exit times (formula proven for 1D):

$$k_{\rm AB}^{\rm ex} = \frac{p_{\rm B}^{\rm eq}}{p_{\rm A}^{\rm eq}\tau_{\rm B}^{\rm ex} + p_{\rm B}^{\rm eq}\tau_{\rm A}^{\rm ex}}$$

relaxation involves both exit and penetration - best approach



Chemical dynamics and rare events

Our approach



Chemical dynamics and rare events

16/23

More formal exit

- looking at relaxation from cell B_i to cell B_j
- $f_{i,k}^e(t)$ probability density that a trajectory initiated at equilibrium in B_i exits this cell toward B_k at time t

exit probability
$$p_{i,k}^e = \int_0^\infty f_{i,k}^e(t) dt$$

mean exit time $\tau_{i,k}^e = \frac{1}{p_{i,k}^e} \int_0^\infty f_{i,k}^e(t) t dt$

for each neighbour B_k of B_i

Chemical dynamics and rare events

17/23

More formal supply

• $f_{i,k}^e(t)$ - probability density that an equilibrium trajectory entering B_i from B_j at time s exits this cell toward B_k at time s + t



for each pair of neighbours of B_i

Chemical dynamics and rare events

18/23

Combining exit times and the TST

• TST rate constant

$$k_{i,j}^{\text{TST}} = \frac{p_{i,j}^e}{\sum_{k\sim i} p_{j,k}^i \tau_{j,k}^i}$$

• Exit time rate constant

$$k_{i,j}^{\text{ex}} = \frac{c_j^e}{c_j^i \tau_{i,j}^e + c_i^e \tau_{j,i}^e}$$

• Relaxation rate constant

$$k_{i,j} = \frac{(p_j^{*,e})^2}{\tau_{j,i}^* - \tau_{i,j}^*}$$

where all 3 terms in the equation are obtained by solving certain sets of linear equations

Chemical dynamics and rare events

19/23

Illustration on Sinai billiards

- non-interacting particles with equal velocities elastically reflect from walls of the billiard and the circle inside
- circle forms a bottleneck transition through the dividing surface is a rare event
- small change in definition of the dividing surface should not cause large change in the rate constants



Toy system: Cluster of Lennard-Jones disks



- search for rotation-permutation isomers
- 10-50 cells placed along the collective coordinate root mean square distance from minimum 2

Chemical dynamics and rare events

21/23

Towards biology



Chemical dynamics and rare events

22/23

Acknowledgements



Chemical dynamics and rare events

23/23